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## Synthesis of a new transition-state analog of the sialyl donor. Inhibition of sialyltransferases

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Abstract—A new class of glycosyltransferase inhibitor has been designed and synthesized. The designed inhibitors 3a/3b provide conformational mimicry of the transition state in sialyltransfer reactions. The key synthetic steps involve a Meinwald rearrangement and a palladium-catalyzed carbonylation reaction. The results of kinetic studies show that 3a/3b exhibit significant inhibition on both 2,3- and 2,6-sialytransferases. © 2001 Published by Elsevier Science Ltd.

Sialic acids play an important role in a variety of biological processes.<sup>1</sup> They are usually attached to the terminal positions of glycoproteins, glycolipids and oligosaccharides. Of more than 100 different sialic acids, N-acetylneuraminic acid (NeuAc) is the most abundant one.<sup>2</sup> The sugar-nucleotide CMP-NeuAc 1 (Fig. 1) is the key substrate for biosynthesis of sialylated glycoconjugates in which 1 is transferred by sialyltransferases to an acceptor hydroxyl group in a variety of substrates including polysialic acids, glycoproteins and gangliosides.<sup>3</sup> These glycosylation patterns constitute important binding determinants for various cellcell interactions which include masking of trypanosomal immunogenicity, viral infection and replication, and cell adhesion.<sup>4</sup> Therefore, development of potent and selective inhibitors of sialyltransferases may be useful in a variety of biochemical applications. However, only a few potent inhibitors of sialyltransferases have so far been developed.<sup>5</sup> We have previously reported the solvolytic behavior and enzymology of 1,

as characterized by kinetic isotope effects which strongly support an oxocarbenium ion-like transition state structure 2 for both solvolysis and the sialyltransferase catalyzed reactions.<sup>6</sup> According to this model, positive charge is accumulated at C-1' and O-1' and the carboxylate group approaches coplanarity with the oxocarbenium ion plane. A common strategy often used to design inhibitors of glycosidases and/or glycosyltransferases is to mimic the electronic charge distribution of a proposed transition state.<sup>7</sup> On the other hand, we noted that in addition to the positive charge and the coplanarity developed in 2, the increased distance between the leaving CMP group and C-1' might be another useful factor for inhibitor design. As a test for this distinct concept, we designed compound 3a, in which the conjugated carboxylate group mimics the oxocarbenium ion coplanarity in transition state 2, and moreover, the CMP group attached to the bicyclic structure is held at a distance to mimic a late transition state for bond cleavage. Here we wish to report the



Figure 1.

Keywords: transition states; inhibitors; glycosyltransferases; sialytransferases; synthesis.

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We envisioned that the bicyclic structure of 3a could be established by Meinwald rearrangement of norbornadiene  $4.^{8}$  With this in mind, we initiated a synthetic sequence, as shown in Scheme 1. Meinwald rearrangement of norbornadiene 4 with *m*-chloroperoxybenzoic acid and subsequent reduction with sodium borohydride gave racemic 6-endo-hydroxymethyl bicyclo-[3.1.0]hex-2-ene. Protection using *tert*-butyldimethylsilvl chloride or *tert*-butyldiphenylsilvl chloride in the presence of imidazole proceeded in 85-90% yields. Hydroboration-oxidation furnished the desired alcohol 5a and its regioisomer 11a in high yields (5a:11a  $\approx$  5:2, by NMR spectroscopy).9 The ratio of diastereomers produced by hydroboration was insensitive to the Oprotecting group. Stereoisomers 5a/11a or 5b/11b could be separated by careful column chromatography. Oxidation of 5a or 5b with chromium trioxide-pyridine gave ketone **6a** (78%) or ketone **6b** (86%), respectively. Alternatively, a mixture of 5b/11b was used for the oxidation reaction to give the desired ketone **6b**, which could be easily separated from its regioisomer by flash column chromatography. Reaction of ketone 6b with trifluoromethanesulfonic anhydride using 2,6-di-tertbutyl-4-methylpyridine as a base afforded vinyl triflate 7 in 69% yield. Palladium-catalyzed carbonylation of 7 in methanol proceeded smoothly to form the unsaturated ester 8 in 80% yield.<sup>10</sup> Deprotection of 8 with tetrabutylammonium fluoride (TBAF) gave the key intermediate 9 in 85% yield.

Cytidine-2',3'-di-*O*-acetyl-4-*N*-acetyl-5'-(*N*,*N*-diisopropyl-2-cyanoethyl)-phosphoramidite was prepared from the triacetyl cytidine<sup>11a</sup> by the literature method.<sup>11c</sup> We found that a more facile synthesis of the triacetyl cytidine could be carried out with the following one-pot sequence. Selective protection of the 5'-hydroxyl group of N-acetyl-cytidine<sup>11b</sup> with 4,4'-dimethoxytrityl chloride in pyridine, followed by acetylation of the 2',3'hydroxyl groups with acetic anhydride in pyridine and deprotection of the 5'-hydroxyl group with 90% aqueous acetic acid, gave the required triacetate in 41% overall yield. The coupling of 9 with the cytidine 5'phosphoramidite in the presence of tetrazole, followed by oxidation with tert-butyl hydroperoxide and basecatalyzed removal of the cyanoethyl group gave a diastereomeric mixture of 10a and 10b. Saponification of 10a/10b with sodium methoxide in methanol-water afforded a mixture of 3a and 3b (1:1 by NMR), which was purified by anion-exchange HPLC (NH4HCO3-MeOH-H<sub>2</sub>O buffer as eluent).<sup>12</sup> Our efforts to separate the diastereoisomers 3a and 3b were unsuccessful, so the mixture of 3a/3b was used for the inhibition studies.

The inhibitory effects of **3a/3b** on rat 2,3- and 2,6-sialytransferases were investigated with the use of radiolabeled [9-<sup>3</sup>H] CMP-NeuAc as a donor substrate. The acceptor sugars employed were lactose (for 2,3-sialytransferase) and lacNac (for 2,6-sialytransferase). The results show that **3a/3b** are competitive inhibitors of both 2,3- and 2,6-sialytransferases, and the  $K_i$ s were estimated to be 10 and 20  $\mu$ M for 2,3- and 2,6-sialytransferases, respectively. These results provide further impetus for development of new analogs that incorporate additional features of the enzyme-bound transition state.

In summary, we report the design and synthesis of a new class of sialyltransferase inhibitors which mimic transition state geometry. The key synthetic steps involve a Meinwald rearrangement and a palladium-catalyzed carbonylation reaction. Despite lacking some of the structural features (e.g. C6–C9 glycerol side chain) of the natural sialyltransferase substrate, CMP-NeuAc, the designed inhibitors 3a/3b showed significant inhibition on both 2,3- and 2,6-sialytransferases.



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Scheme 1. Reagents and conditions: (a) (i) m-CPBA; (ii) NaBH<sub>4</sub>; (iii) TBDMSCl or TBDPSCl, imidazole. (b) (i) BH<sub>3</sub>·SMe<sub>2</sub>; (ii) H<sub>2</sub>O<sub>2</sub>, NaOH, 90–95%. (c) CrO<sub>3</sub>, pyridine, 78% for **6a**; 86% for **6b**. (d) Tf<sub>2</sub>O, 2,6-di-*tert*-butyl-4-methylpyridine, 69%. (e) Pd(OAc)<sub>2</sub>, PPh<sub>3</sub>, Et<sub>3</sub>N, MeOH, CO, DMF, rt, 80%. (f) TBAF, THF, 85%. (g) (i) Cytidine-2',3'-di-O-acetyl-4-N-acetyl-5'-(N,N-diisopropyl-2-cyanoethyl)-phosphoramidite, 1*H*-tetrazole, CH<sub>2</sub>Cl<sub>2</sub>; (ii) *t*-BuOOH; (iii) Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, 51% for three steps. (h) (i) NaOMe, MeOH–H<sub>2</sub>O; (ii) HPLC (MonoQ, NH<sub>4</sub>HCO<sub>3</sub>–MeOH–H<sub>2</sub>O); (iii) desalting with Amberlite IR-120, 30%.

We expect that analogs of 3a/3b will exhibit more potent and selective inhibition on sialyltransferases, and may be extended to other glycosyltransferases.

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- 12. Selected data for **9**: <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>Cl)  $\delta$  6.75 (m, 1H), 3.71 (s, 3H), 3.53 (m, 1H), 3.36 (m, 1H), 2.76 (m, 1H), 2.45 (m, 1H), 2.20 (m, 1H), 1.89 (m, 1H), 1.48 (m, 1H), 1.38 (bs, 1H); <sup>13</sup>C NMR (75 MHz, CD<sub>3</sub>Cl)  $\delta$  164.87, 141.61, 135.68, 56.31, 51.39, 31.14, 28.88, 24.10, 21.38; HRMS calcd for C<sub>9</sub>H<sub>13</sub>O<sub>3</sub> (M<sup>+</sup>+H): 169.0865; found: 169.090. For **3a/3b**: <sup>1</sup>H NMR (300 MHz, D<sub>2</sub>O)  $\delta$  7.75 (d, *J*=7.62 Hz, 1H), 6.29 (bs, 1H), 5.88 (d, *J*=7.62 Hz, 1H), 5.75 (m, 1H), 4.45 (m, 3H), 3.93 (m, 1H), 3.83 (m, 1H), 3.58 (m, 1H), 3.44 (m, 1H), 2.45 (m, 1H), 2.20–2.18 (m, 2H), 1.69 (m, 1H), 1.28 (m, 1H); <sup>13</sup>C NMR (75 MHz, D<sub>2</sub>O)  $\delta$  181.87, 173.91, 166.47, 158.12, 142.10, 141.61, 137.61, 96.82, 89.63, 89.52, 83.01, 74.46, 69.70, 64.46, 61.68, 31.96, 28.83, 23.65, 21.01; HRMS calcd for C<sub>17</sub>H<sub>21</sub>N<sub>3</sub>O<sub>10</sub> P (M<sup>+</sup>–2Na<sup>+</sup>+H): 458.0965; found: 458.087.